Quiz 1

- 1. Sketch the $2p_x$ and $2p_y$ orbitals (around the same nucleus, please).
- 2.
- a. Write the electron configuration of carbon.
- b. Are the p electrons in the same orbital? Why or why not?
- 3. What element is represented by the following ground state electron configuration?

$$1s^22s^22p^63s^23p^63d^2$$

- 4. Answer the following questions about NH₃ (ammonia):
 - a. What type of bond is formed between nitrogen and hydrogen?
 - b. Draw a Lewis structure for ammonia.
 - c. What shape is the smallest unit of ammonia? (Name the shape and draw a structure.)